

Physiological responses of *Ulva pertusa* and *U. armoricana* to copper exposure

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Abstract

A comparative study of copper (Cu) toxicity and tolerance in two species of *Ulva* from Korea, the native *Ulva pertusa* and alien *Ulva armoricana*, was conducted by examining the effects on growth, pigmentation, chlorophyll fluorescence, antioxidant capacity and nitrate reductase activity. Toxic effects of Cu were less expressed in *U. armoricana* than in *U. pertusa*. At lower concentrations (25–50 $\mu\text{g L}^{-1}$), exposure to Cu did not affect thallus growth of *U. armoricana*, whilst growth was significantly reduced in *U. pertusa*. An increase in chlorophyll concentrations was observed in *U. armoricana* exposed up to 100 $\mu\text{g L}^{-1}$, whereas Cu caused a significant chlorophyll reduction in *U. pertusa*. Chlorophyll *b* was reduced to a lesser extent than chlorophyll *a* by higher Cu concentrations. In *U. armoricana*, the maximum efficiency of photosystem II, minimum fluorescence, maximum electron transport rate and non-photochemical quenching were unaffected by Cu except at the highest concentration tested. *U. pertusa* showed a significant decrease in those parameters at much lower Cu concentrations. It was notable that in this alga the maximum efficiency of photosystem II was reduced at higher Cu concentrations than relative electron transport rate. Elevated concentrations of Cu induced a strong activation of antioxidant activity in *U. armoricana*, whereas the generation of high levels of reactive oxygen species probably decreased the non-enzymatic antioxidant defense system in *U. pertusa*. An increase in the nitrate reductase activity of *U. armoricana* at 50–100 $\mu\text{g L}^{-1}$ Cu coincided with the increase in chlorophyll contents, whereas *U. pertusa* showed a significant decrease at the higher Cu concentration. Differences in the sensitivity of the two species of *Ulva* to Cu may influence their competitive interactions in Korean coastal waters experiencing temporal increases in the loading of heavy metals.

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1. Introduction

Species of the green seaweed genus *Ulva* are generally considered to be tolerant of a wide range of environmental stressors, including metals (Correa et al., 1996; Fletcher, 1996) and because of their high accumulation capacity for metals they are

used as bioindicators of metal pollution (Haritonidis and Malea, 1999; Lee and Wang, 2001). However, the results obtained from recent studies on *U. pertusa* suggest that this species might be more sensitive to toxic chemicals, including metals, than other *Ulva* species and some freshwater microalgae and aquatic plants (Han and Choi, 2005; Han et al., 2007). This conclusion is derived from comparisons with published data (e.g. EC₅₀s) and does not take account of differences in the culture conditions used in the various studies, which can influence the toxicity of metals (Brown and Newman, 2003). Therefore, to establish whether related species have evolved differential tolerance to metals, comparative studies using material collected from the same location need to be undertaken. Despite much published data on the impacts of metals on macroalgae very few studies have investigated directly inter-specific differences in response to metal exposure.

Abbreviations: chl, chlorophyll; DPPH, α,α -diphenyl- β -picrylhydrazyl; F_v/F_m , maximum efficiency of PS II; F_v , variable chl fluorescence; F_m , the maximum fluorescence yield of dark-adapted samples; F_0 , the minimum fluorescence yield of dark-adapted samples; ETR_{max}, maximum electron transport rate; NR, nitrate reductase; NPQ, non-photochemical quenching; PS II, photosystem II; PPFD, photosynthetic photon flux density; ROS, reactive oxygen species.

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U. pertusa Kjellman is native to Korea and forms extensive mats in shallow coastal waters due to its fast rate of growth and high reproductive capacity (Han et al., 2003). This seaweed plays an important role as a primary producer, contributing directly and indirectly to higher trophic levels, and provides habitat and nursery grounds for fish and invertebrates. Recently, extensive populations of the related alien species, *Ulva armoricana* Dion et al. have been recorded from the eastern coasts of Korea, and found growing in the same communities as *U. pertusa* (Han et al., 2006). The introduction of any new seaweed raises concerns about the potential risks to the indigenous biota by, for example, pre-empting space and resources for related species, reducing food and shelter for consumers, threatening biodiversity and, more generally, disturbing the dynamics of near-shore ecosystems (Carlton, 2000; Occhipinti-Ambrogi and Savini, 2003). In the case of *U. armoricana* there is evidence for environmental problems associated with its introduction into French coastal waters including damage to shellfish cultivation through fouling, production of large-scale green tides and catastrophic decomposition of the algal mats resulting in dystrophic conditions (Dion and Le Bozec, 1996). Compared with other *Ulva* species, *U. armoricana* displays a high photosynthetic potential, nutrient uptake performance and wide temperature optimum (Dion and Le Bozec, 1996; Dion et al., 1998), but its response to metal pollution has not previously been investigated. As both species co-occur in Korean waters impacted to varying degrees by anthropogenically derived chemical pollutants, there is an opportunity to investigate their responses to metal-pollution and to compare their levels of resistance.

Although copper (Cu) is essential for the metabolism of eukaryotic organisms it is generally considered to be one of the most toxic metals (Lobban and Harrison, 1994; Gledhill et al., 1997). At concentrations beyond a certain threshold, Cu induces phytotoxic symptoms in algae such as: inhibition of growth, photosynthesis and fertility, disruption of development, chlorosis and damage to membranes (Gledhill et al., 1997; Brown and Newman, 2003; Nielsen et al., 2003a,b; Eklund and Kautsky, 2003). In off-shore seawater, reported values for total Cu concentration vary from 0.002 to 6 $\mu\text{g L}^{-1}$, but the concentration may increase several fold in coastal waters affected by waste derived from e.g., industrial activities, sewage, antifouling paints and pesticides (Kim and Han, 1999; Brown and Newman, 2003). In Korean coastal waters and estuaries dissolved Cu concentrations range between 4 and 14 $\mu\text{g L}^{-1}$ (Korea Maritime Ministry), and these levels are likely to increase with time as a result of industrial growth and population increase (Lee et al., 1998).

To assess the relative sensitivities of the two species of *Ulva* to Cu, several end-points at different levels of biological organization were used. Such an approach allows for a more detailed evaluation of the response to an environmental stressor than using a single physiological parameter, and can provide information on the mechanism(s) of resistance to that stressor (Brown and Newman, 2003). Five parameters were selected: growth, pigmentation, chlorophyll fluorescence, antioxidant and nitrate reductase activities, all of which are known to be affected, to varying degrees, by Cu in other plants and algae (e.g. Cid et al., 1995; Ouzounidou et al., 1997; Brown and Newman, 2003).

2. Materials and methods

2.1. Culture of plants

U. pertusa and *U. armoricana* were collected from sites near Ahnin on the well-conserved eastern coast of Korea (37.4°N, 129.1°E). Unialgal stock cultures were maintained in artificial seawater medium, incubated at 15 °C and 10–15 $\mu\text{mol photons m}^{-2} \text{s}^{-1}$ of white fluorescent light (FL400, Kum-Ho, Seoul, South Korea) on a 12-h photoperiod. The medium was prepared by dissolving commercial sea salts (Coralife, Energy Savers, California, USA) in deionized water to a concentration of 35‰, to which was added 1 mM KNO_3 and 0.1 mM K_2HPO_4 as nutrients.

Algal disks (\varnothing 6 mm) were cut from the middle region of healthy thalli and placed in flasks (15 ml volume per well) with medium containing different concentrations of Cu. The medium was gently bubbled with air. A stock solution of copper chloride (CuCl_2) was prepared from a concentrated standard in deionized water acidified with 1N hydrochloric acid (HCl) (Junsei, Tokyo, Japan). Appropriate volumes of the stock solution were added so that the final concentrations were in the range 25–250 $\mu\text{g L}^{-1}$ Cu. Cultures were established under optimal conditions for growth i.e.: 15 °C, 80–100 $\mu\text{mol photons m}^{-2} \text{s}^{-1}$ of white fluorescent light, 12:12-h LD photoperiod and salinity of 35‰. Controls consisted of the artificial seawater medium without added Cu. There were four replicates per treatment.

2.2. Growth

Following exposure to experimental treatments for 3 days, the algal samples were harvested to determine changes to disk area, measured using a computer-assisted image analyzer (MV200, Samsung, Seoul, Korea). Relative growth rates (RGR) were calculated using the equation:

$$\text{RGR (\%/day)} = \frac{\ln A_t - \ln A_0}{t} \times 100\%$$

where A_0 and A_t are the areas of disks at time zero and after t days, respectively.

2.3. Chlorophyll concentrations

Chlorophyll concentrations were determined using a Specord spectrophotometer (S10, Zeiss) after extraction of disks in 100% methanol for at least 24 h in the dark at 4 °C. For chlorophyll *a* and *b* contents the methanolic extracts were measured at 666 nm (chlorophyll *a*; chl *a*) and at 653 nm (chlorophyll *b*; chl *b*) using the equations outlined in Hipkins and Baker (1986). The chl *a/b* ratio was also calculated for all disks.

2.4. Chlorophyll fluorescence

Chlorophyll fluorescence was measured using a pulse amplitude modulated (PAM) fluorometer (Diving-PAM, Walz,

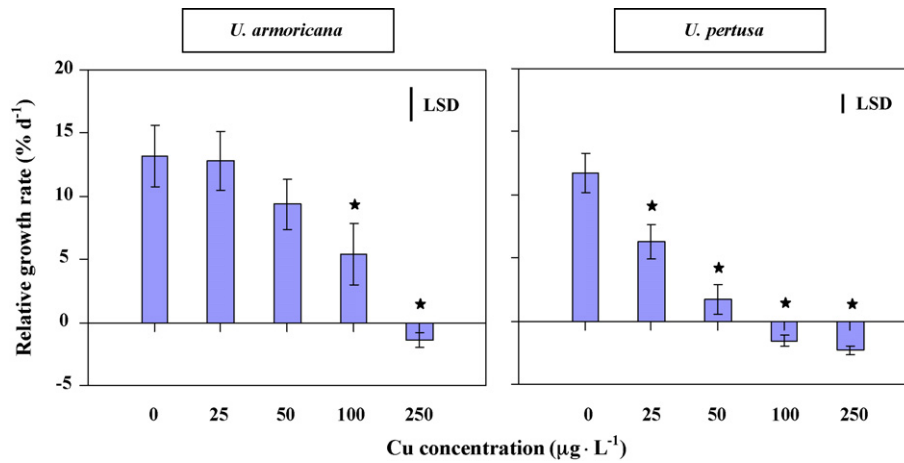


Fig. 1. Daily relative growth rate of *Uva armoricana* and *U. pertusa* after a 3-day exposure to a range of Cu concentrations. Mean \pm 95% confidence interval ($n=11$) are shown. Values with an asterisk are significantly different at $P<0.05$ from those of the controls.

Effeltrich, Germany). Samples were initially dark adapted for 15 min before chl fluorescence readings were taken. F_m , the maximum fluorescence yield of dark-adapted samples and F_0 , the initial fluorescence yield, were recorded. The maximum quantum yield of PSII in the dark-adapted state is expressed as the ratio of variable to maximal chl fluorescence (F_v/F_m), derived from $(F_m - F_0)/F_m$.

Rapid light curves were measured using 10 s pulses of actinic light increased stepwise from 0 to 1517 $\mu\text{mol photons m}^{-2} \text{s}^{-1}$ (Schreiber, 2004). Relative electron transport rates (rel.ETR) were calculated by multiplying the effective quantum yield ($\Phi_{\text{PSII}} = F'_m - F/F'_m$, where F'_m is maximum light-acclimated fluorescence yield and F is light-acclimated fluorescence yield) by photosynthetic photon flux density (PPFD) and plotted against PPFD. The ETR is relative because the absorbance of the thallus was not measured. Maximum electron transport rate (ETR_{max}) was derived from the hyperbolic tangent formulation, $\text{rel.ETR} = \text{ETR}_{\text{max}} \tanh(\alpha I / \text{ETR}_{\text{max}})$, where α indicates electron transport rate under light-limited conditions, adapted from Jassby and Platt (1976). Alterations in chl fluorescence due to changes in non-photochemical quenching (NPQ) were calculated from $(F_m - F'_m)/F'_m$ (Maxwell and Johnson, 2000).

2.5. Antioxidant activity

Non-enzymatic antioxidant activity of disks was measured by DPPH (α, α -diphenyl- β -picrylhydrazyl) scavenging activity. Thallus disks were homogenized in 2 ml of absolute ethanol with a mortar and pestle and the extracts centrifuged at 10,000 rpm for 15 min. A 0.5-ml aliquot of the supernatant was mixed with a 0.5-mM DPPH ethanol solution (0.25 ml) and 100 mM of acetate buffer (pH 5.5, 0.5 ml). After incubating in the dark for 30 min, the absorbance of the mixture was measured at 517 nm to determine DPPH scavenging activity (Abe et al., 1998). Activity is expressed as $(A_{\text{control}} - A_{\text{treated}})/A_{\text{control}} \times 100\%$, where A_{control} and A_{treated} are the absorbances at 517 nm of control and copper-treated samples.

2.6. Nitrate reductase (NR) activity

NR activity was assayed as described by Jun and Kim (1989) with some modifications. After a 72-h culture period under different Cu treatment conditions (50 and 100 $\mu\text{g L}^{-1}$), disks were harvested and weighed before each was transferred to a flask containing 20 ml of fresh medium with 20 mM sodium nitrate, 4% n-propanol and chloramphenicol (2 $\mu\text{g L}^{-1}$) for 3-h incubation in the dark. NR activity was determined by incubating 1 ml supernatant with 3N HCl and 0.02% *N*-1-naphthyl-ethylenediamine dihydrochloride, at room temperature for 20 min. The amount of nitrite produced was determined by measuring the absorbance at 540 nm. One unit of NR activity is defined as that which produced 1 mol $\text{NO}_2^{-1} \text{h}^{-1}$.

2.7. Statistical analysis

One-way analysis of variance (ANOVA) was performed to confirm significant differences in response. Multiple comparison test by the least significance difference (LSD) was then carried out to find out significant differences at $P=0.05$ in response from controls.

3. Results

3.1. Effects of copper on growth

Toxic effects of Cu on growth were strongly dependent on species. The mean RGR of *U. armoricana* was not significantly affected by 25–50 $\mu\text{g L}^{-1}$ Cu but at 100 $\mu\text{g L}^{-1}$ Cu the mean RGR was reduced to 41% of the control and at 250 $\mu\text{g L}^{-1}$ the RGR was negative ($-1.4\%/ \text{day}$), indicating a shrinkage of the disks (Fig. 1). *U. pertusa* was less tolerant to Cu with a 43% reduction of growth even at the lowest test concentration (25 $\mu\text{g L}^{-1}$) and negative mean RGRs at 100 and 250 $\mu\text{g L}^{-1}$ Cu (Fig. 1).

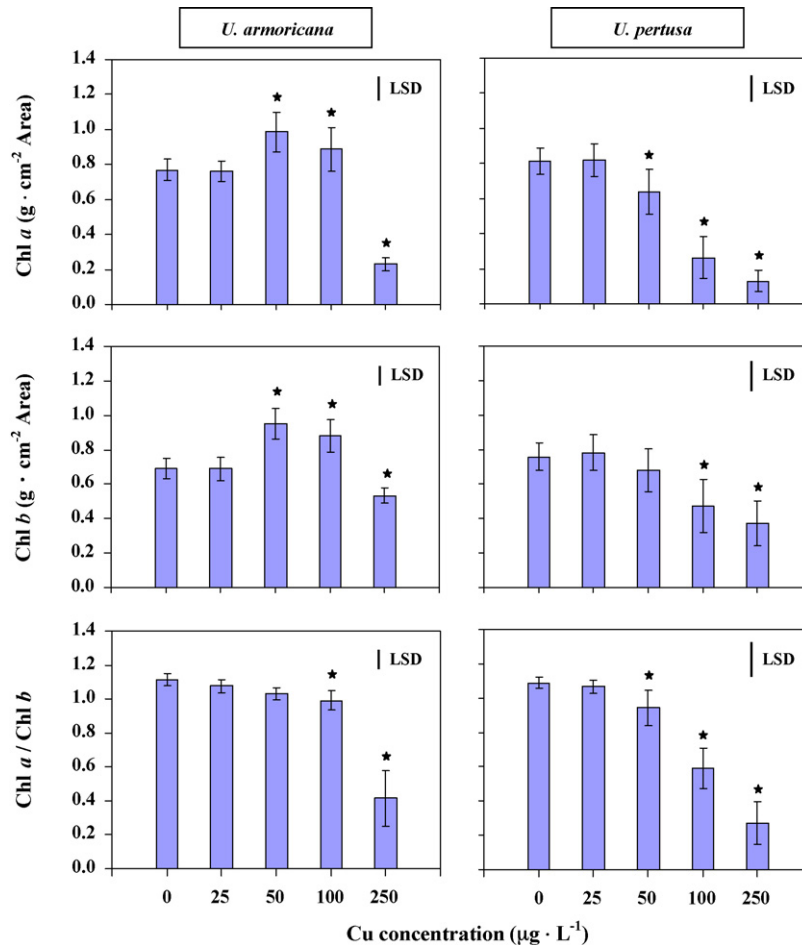


Fig. 2. Chlorophyll concentrations of *Ulva armoricana* and *U. pertusa* after a 3-day exposure to a range of Cu concentrations. Mean \pm 95% confidence interval ($n=11$) are shown. Values with an asterisk are significantly different at $P<0.05$ from those of the controls.

3.2. Effects of copper on chlorophyll concentrations

Fig. 2a and b show the changes in the concentrations of chl *a* and *b* with increasing Cu for both species. When exposed to $50 \mu\text{g L}^{-1}$ Cu the concentrations of both pigments, relative to the controls, increased by 29–36.2% in *U. armoricana* but decreased by 14–23% in *U. pertusa*. At $100 \mu\text{g L}^{-1}$ Cu, *U. armoricana* still exhibited 5–26% higher concentrations of chl *a* and *b* than the controls, whereas the same treatment reduced the chlorophyll contents of *U. pertusa* by 42–70%. At the highest tested concentration of Cu, the two pigments were significantly reduced in both species, although to lesser extent in *U. armoricana*. For both species, reductions in chl *b* concentrations at 50– $250 \mu\text{g L}^{-1}$ Cu were less than for chl *a*. The ratio of chl *a/b* (Fig. 2c) remained stable in the Cu concentration range between 25 and $50 \mu\text{g L}^{-1}$ but significantly decreased at 100 (11%) and $250 \mu\text{g L}^{-1}$ (63%) for *U. armoricana* whereas in *U. pertusa* there was a significant decrease in the ratio at concentrations greater than $25 \mu\text{g L}^{-1}$.

3.3. Effects of copper on photosynthetic activity

A significant decline in the maximum quantum yield of PS II, as measured by F_v/F_m , was apparent at $250 \mu\text{g L}^{-1}$ Cu for *U. armoricana* and $100 \mu\text{g L}^{-1}$ for *U. pertusa* (Fig. 3). *U.*

armoricana exhibited a relatively stable F_0 up to $100 \mu\text{g L}^{-1}$ Cu but at $250 \mu\text{g L}^{-1}$ values declined significantly. In *U. pertusa*, F_0 was unaffected by Cu concentrations 25– $50 \mu\text{g L}^{-1}$, but values were significantly lower at 100 and $250 \mu\text{g L}^{-1}$ Cu.

A significant reduction of ETR_{max} occurred at $250 \mu\text{g L}^{-1}$ Cu for *U. armoricana* while even the lowest tested Cu concentration caused 19% reduction in *U. pertusa*, followed by further significant reductions at higher concentrations.

In *U. armoricana*, values of NPQ remained unaltered in disks exposed to concentrations less than $250 \mu\text{g L}^{-1}$ Cu, whereas in *U. pertusa* NPQ decreased significantly from control values at all external concentrations of Cu. In both species NPQ values fell to just above zero at $250 \mu\text{g L}^{-1}$.

3.4. Effects of copper on non-enzymatic antioxidant activity

In *U. armoricana*, exposure to concentrations of 25 to $100 \mu\text{g L}^{-1}$ Cu resulted in a stimulation of antioxidant activity, reaching its maximum at $100 \mu\text{g L}^{-1}$ (Fig. 4). In contrast, the activities of antioxidants showed a decreasing tendency in *U. pertusa* exposed to elevated Cu. At $250 \mu\text{g L}^{-1}$ Cu antioxidant activities of both species were significantly reduced to 20–30% of control values.

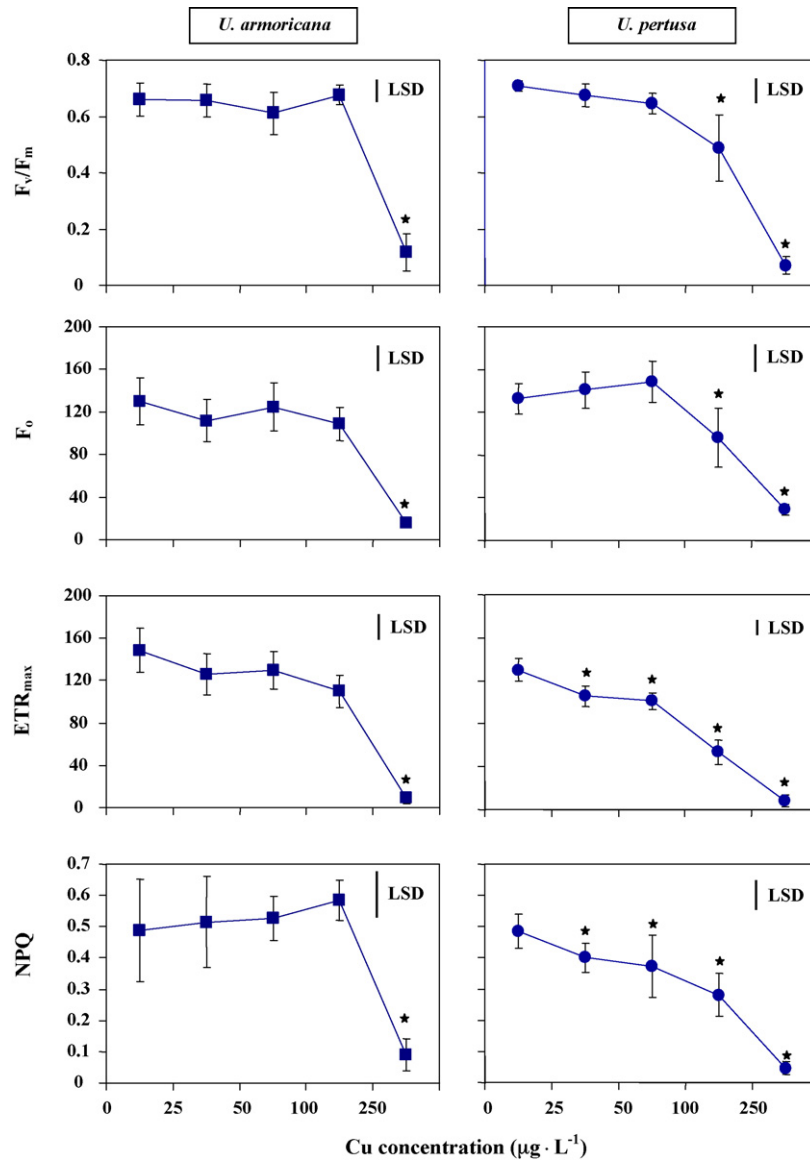


Fig. 3. Chlorophyll fluorescence parameters, F_v/F_m , F_0 , ETR_{max} , and NPQ, measured on *Ulva armoricana* and *U. pertusa* after a 3-day exposure to a range of Cu concentrations. Mean \pm 95% confidence interval ($n = 11$) are shown. Values with an asterisk are significantly different at $P < 0.05$ from those of the controls.

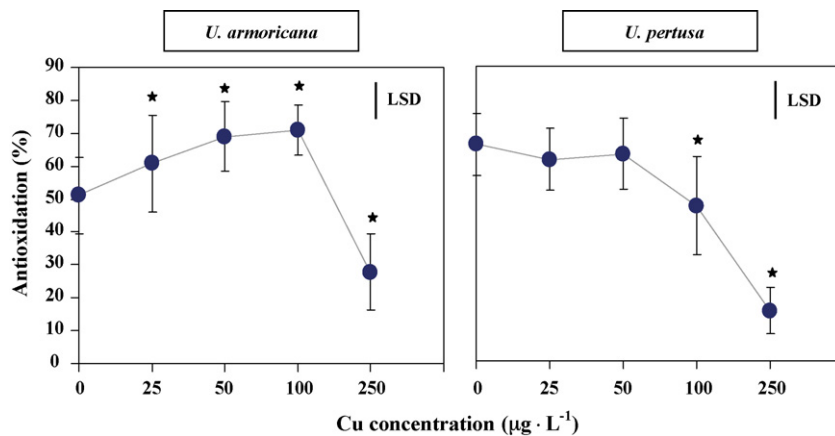


Fig. 4. Non-enzymatic antioxidant activity of *Ulva armoricana* and *U. pertusa* after a 3-day exposure to a range of Cu concentrations. Mean \pm 95% confidence interval ($n = 11$) are shown. Values with an asterisk are significantly different at $P < 0.05$ from those of the controls.

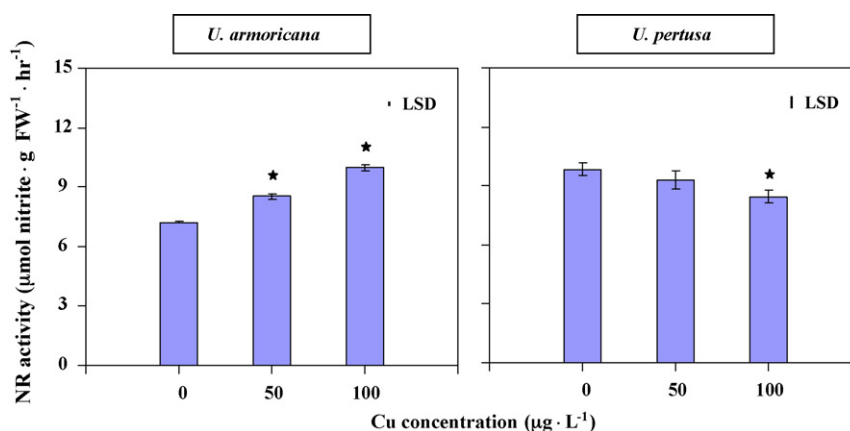


Fig. 5. NR activity of *Ulva armoricana* and *U. pertusa* after a 3-day exposure to 50–100 $\mu\text{g L}^{-1}$ Cu. Mean \pm 95% confidence interval ($n=11$) are shown. Values with an asterisk are significantly different at $P < 0.05$ from those of the controls.

3.5. Effects of copper on nitrate reductase activity

Copper induced differential effects on NR activity of the *Ulva* species. *U. armoricana* showed 18 and 38% increase in NR activity relative to the controls after exposure to 50 and 100 $\mu\text{g L}^{-1}$ Cu, respectively (Fig. 5). In contrast, a significant decrease in NR activity (14%) was observed in *U. pertusa* exposed to 100 $\mu\text{g L}^{-1}$ Cu.

4. Discussion

Differences in the response of the two species of *Ulva* to Cu exposure have been identified. The results obtained from measurements at different levels of biological organization show that *U. armoricana* is more tolerant to copper than *U. pertusa* within the concentration range used in the current study. Of the various endpoints measured, growth was found to be the most sensitive. The threshold concentration that resulted in a significant reduction in relative growth rate (RGR) was 25 $\mu\text{g L}^{-1}$ for *U. pertusa* compared with 100 $\mu\text{g L}^{-1}$ for *U. armoricana*. Studies on other *Ulva* species have also reported that significant reductions in growth occur at concentrations greater than 100 $\mu\text{g L}^{-1}$; for example, in *U. compressa* (L.) Nees von Esenbeck growth inhibition occurred in the range 114–6354 $\mu\text{g L}^{-1}$ Cu (Reed and Moffat, 1983; Correa et al., 1996) and in *U. fasciata* Delile 50% growth inhibition was recorded at 850 $\mu\text{g L}^{-1}$ Cu (Lee et al., 2005). On exposure to 250 $\mu\text{g L}^{-1}$ shrinkage of thallus disks was observed in both species, a finding that has been reported previously in the red seaweed *Gracilariopsis longissima* (S.G. Gmelin) Steentoft, LM Irvine and Farnham and the brown seaweed *Fucus vesiculosus* L. exposed to similar concentrations of Cu (Bryan and Gibbs, 1983; Brown and Newman, 2003). Copper-induced interference with cell division and/or expansion has been proposed as a possible reason for the observed reduction in growth (e.g. Stauber and Florence, 1987). This may be linked to a decrease turgor and/or a change in cell wall elasticity due to copper toxicity (Brown and Newman, 2003).

Uncoupling of growth from photosynthetic electron transport at elevated concentrations of Cu was evident in both species of

Ulva. An increase in Cu from 25 to 50 $\mu\text{g L}^{-1}$ caused a significant reduction in growth in *U. pertusa* but had no apparent effect on photosynthetic electron transport, while for *U. armoricana* a similar response was observed only at higher concentrations (50–100 $\mu\text{g L}^{-1}$). These results imply that the initial site of toxic action does not occur in pathways associated with O_2 evolution. Similar conclusions have been reached for several microalgae and seaweeds (e.g. Lumsden and Florence, 1983; Nalewajko and Olaveson, 1995; Brown and Newman, 2003). For example, Brown and Newman (2003) reported that while growth rates of the red seaweed *G. longissima* were reduced significantly at 12.5 $\mu\text{g L}^{-1}$, no observable changes in photosynthesis were evident until 250 $\mu\text{g L}^{-1}$ Cu. They concluded that this uncoupling of growth from photosynthesis at low concentrations of Cu was most likely due to the energy captured by the light reactions being utilized to combat the toxic effects of Cu through induction of protective mechanisms. Further support for this view is provided by the results obtained in the current study.

Photosynthetic activity of *U. pertusa* and *U. armoricana*, as determined from the chlorophyll fluorescence parameters, was not significantly affected until the seaweeds were exposed to 100 and 250 $\mu\text{g L}^{-1}$ Cu, respectively. The decrease in activity was related to changes in the chlorophyll content of disks. A reduction in chlorophyll, due to either a decrease in biosynthesis or increase in the rate of degradation, is a commonly observed symptom of toxicity in algae (Gledhill et al., 1997). Copper is known to regulate the pigment content and composition of the pigment-protein complexes of PS II (Prasad and Strzalka, 1999). For *U. pertusa*, exposure to 50 $\mu\text{g L}^{-1}$ was found to significantly lower the concentration of chl *a*, and the chl *a/b* ratio, with further progressive reductions at the higher concentrations, whereas the concentration of chl *b* was only affected at higher concentrations of Cu. The greater susceptibility of chlorophyll *a* to Cu exposure has also been observed in a freshwater macrophyte, *Ceratophyllum demersum* L. (Coontail) and some lichens (Chettri et al., 1998; Rama Devi and Prasad, 1998). Of note was the lack of change in F_v/F_m and F_0 in *U. pertusa* exposed to Cu concentrations less than 100 $\mu\text{g L}^{-1}$ although chl *a* declined significantly. This disparity might imply that the chl *a* affected by exposure to 50 $\mu\text{g L}^{-1}$ Cu was mostly a component of the

PS I since PS I consists almost entirely of chl *a* while PAM fluorescence probes only PS II.

In contrast, there was an increase in chlorophyll concentrations of *U. armoricana* exposed to 50–100 $\mu\text{g L}^{-1}$ of Cu. Moreover, there was no decrease in either F_v/F_m or F_0 at 100 $\mu\text{g L}^{-1}$, although the RGR was reduced, albeit not to the same extent as in *U. pertusa*. These findings provide further support for a trade-off between energetic resources being used for pigment biosynthesis at the expense of growth. Such a stimulatory effect of excess of Cu on chlorophyll accumulation has previously been observed in copper-tolerant higher plants (Baszyński et al., 1982; Maksymiec and Baszyński, 1996). More generally, significant increases in chlorophyll have been found to occur in response to a range of environmental stresses and are associated with stress resistance (Zhang et al., 2005). For example, in *U. rigida* C. Agardh increases in chl *a* and *b* were considered to be an efficient protective mechanism against ultraviolet-B radiation (UV-B) (Altamirano et al., 2000). It has been noticed in the aquatic plant, *Lemma gibba* that copper and UV radiation elicit similar stress responses (Babu et al., 2003). In both species, the significant decline in F_0 and F_v/F_m at concentrations greater than 100 $\mu\text{g L}^{-1}$ Cu coincided with negative RGRs and are a result of marked reductions in light absorption and PS II efficiency, respectively due to the destruction of chlorophyll molecules (Capsi et al., 1999).

Relative ETR provides an approximation of the rate of electrons pumped via PS II into the photosynthetic chain (Schreiber, 2004). Copper has been shown to inhibit photosynthetic electron transport through damage to both the donor and acceptor sides of PS II (Pätsikkä et al., 1998). In principle, blockage of electron transport by Cu could lead to a decrease in F_v/F_m (Karukstis, 1991). In *U. armoricana*, changes in ETR_{max} paralleled those of F_v/F_m , whereas in *U. pertusa*, ETR_{max} showed a significant decrease at lower Cu concentrations than did F_v/F_m . The disparity between the responses of F_v/F_m and ETR_{max} reflects inhibition downstream of PS II, suggesting that the down-regulation of PS II electron transport is a result of inhibition of Calvin cycle enzymes and Rubisco (Yu et al., 2002).

Transition metals such as Cu catalyze the formation of hydroxyl radicals ($\bullet\text{OH}$) from the non-enzymatic chemical reaction (Haber-Weiss reaction) between superoxide ($\text{O}_2^{\bullet-}$) and H_2O_2 (Winterbourn, 1982). Therefore, the presence of excess Cu can promote oxidative damage by increasing the cellular concentrations of reactive oxygen species (ROS) (Pinto et al., 2003). In chloroplasts, ROS cause inactivation and degradation of Rubisco and other components of the Calvin cycle (Pätsikkä et al., 1998), and lipid peroxidation, resulting in the disruption to photosynthetic pigments (Pätsikkä et al., 2002). The tolerance of photosynthetic organisms to such stress factors is associated with the antioxidant capacity, as the increased levels of antioxidant constituents can prevent stress damage. Copper is known to induce the activity of antioxidants such as ascorbate and glutathione via the ascorbate-glutathione cycle, which plays an important role in detoxification of certain ROS (Panda et al., 2003). However, the responses of antioxidant and enzyme activities in photosynthetic organisms exposed to Cu have not been fully resolved, with their stimulation and inhibition both being

reported (Mazhoudi et al., 1997). In the present study, increase in non-enzymatic antioxidant activity of *U. armoricana* after exposure to sublethal levels of Cu may alleviate the deleterious effects of copper while decreased antioxidant capacity in *U. pertusa* may have resulted from Cu-induced decline of antioxidant activity (Gallego et al., 1996). It is hypothesized, therefore, that the antioxidant capacity of *U. pertusa* was overloaded by the high levels of ROS produced in copper-treated thallus of this species, although further study is required to confirm this.

While NPQ was unaltered in *U. armoricana* by copper treatments up to 100 $\mu\text{g L}^{-1}$, it declined at 25 $\mu\text{g L}^{-1}$ in *U. pertusa*. The copper-induced decrease in NPQ might be explained by a decrease in the rate of electron transport in the pathway $\text{P680} \rightarrow \text{pheophytin} \rightarrow \text{Q}_A \rightarrow \text{Q}_B$ or to an increase in the rate of the back reaction, $\text{Q}_A^- \rightarrow \text{P680}^+$ (Horton and Bowyer, 1990).

Nitrate reductase is one of the most important enzymes of any biosynthetic pathway in the algae and is involved in the assimilation of nitrate, the primary nitrogen source (Berges, 1997). In macroalgae, NR activity is known to be regulated by external factors such as light and nitrogen availability, and by an internal biological clock (Davison and Stewart, 1984; Lopes et al., 1997; Chow et al., 2004). It is interesting to note that in *U. armoricana* the activity of NR increased after exposure to 50–100 $\mu\text{g L}^{-1}$ Cu, compared with the controls, and was coincident with increases in the concentration of chlorophyll and the reduction in RGR. As far as we are aware, Cu-induced stimulation of NR activity has never been reported before. In higher plants, it has been observed that some heavy metals cause *de novo* synthesis of NR isoenzymes (Panda and Patra, 1998). By comparison the Cu-induced reduction in NR activity in *U. pertusa* concurs with other reports of the toxic effects of elevated levels of Cu on NR activity in microalgae (Rai et al., 1994; Tripathi et al., 2004). There are three possible mechanisms by which Cu could have inhibited NR activity: (1) direct modification of thiol groups, (2) oxidative stress occurring through interference with the electron flux from PS II to PS I, and (3) reduced photosynthesis (González-Moreno et al., 1997). Given that ETR_{max} was reduced at 50 $\mu\text{g L}^{-1}$ Cu while NR activity remained constant excludes option (3) as an explanation for the reduced NR activity in *U. pertusa*. The reduction in both DPPH and NR activity at 100 $\mu\text{g L}^{-1}$ Cu would suggest that oxidative damage, as a result of Cu toxicity, is the most likely explanation for these observations.

5. Conclusion

Ulva species are generally considered to be tolerant of the effects of heavy metals compared with many other macroalgae (Haritonidis and Malea, 1999; Lee and Wang, 2001). For the first time we have identified two species of *Ulva*, growing together on the east coasts of Korea, that differ in their sensitivity to Cu. The alien species *U. armoricana* appears to be more tolerant to Cu exposure than the native *U. pertusa* in terms of growth, pigmentation, photosynthesis, antioxidant capacity and NR activity. The observed uncoupling of growth from photosynthetic activity at higher concentrations of Cu provided sufficient energy for the induction of protective measurements, including the synthe-

sis of chlorophyll and antioxidants, to combat Cu-toxicity in *U. armoricana*. This greater level of tolerance to Cu, and potentially other metals, displayed by *U. armoricana*, together with its ability to rapidly colonize disturbed habitats, could result in the native *U. pertusa* being out-competed and displaced and, as a consequence, the biodiversity and functioning of near-shore marine ecosystems in Korea being modified.

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