

Approximated expression of the hygroscopic growth factor for polydispersed aerosols

Chang H. Jung^{a,*}, Young Jun Yoon^b, Junshik Um^c, Seoung Soo Lee^d,
Kyung Man Han^e, Hye Jung Shin^f, Ji Yi Lee^g, Yong Pyo Kim^h

^a Department of Health Management, Kyungin Women's University, Incheon, 21041, Republic of Korea

^b Korea Polar Research Institute, Incheon, 21990, Republic of Korea

^c Department of Atmospheric Sciences, Pusan National University, Busan, 46241, Republic of Korea

^d Earth System Science Interdisciplinary Center, University of Maryland, College Park, MD, 20740, USA

^e School of Earth Sciences and Environmental Engineering, Gwangju Institute of Science and Technology, Gwangju, 61005, Republic of Korea

^f Climate and Air Quality Research Department, National Institute of Environmental Research, Incheon, 22689, Republic of Korea

^g Department of Environmental Science and Engineering, Ewha Womans University, Seoul, 03760, Republic of Korea

^h Department of Chemical Engineering and Materials Science, Ewha Womans University, Seoul, 03760, Republic of Korea

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ABSTRACT

Hygroscopic growth of aerosols plays an important role in the characterization of atmospheric aerosols. Physico-chemical and optical properties of aerosols are dependent on relative humidity (RH) as well as on their size and composition. In this study, scattering enhancement factors, $f(RH)$, for polydispersed aerosols were approximated. $f(RH)$ of ammonium sulfate and ammonium nitrate (AS) and NaCl aerosols (ss) were considered under the assumption of externally mixed aerosols. $f(RH)$ was calculated using the Mie theory for polydispersed aerosols at a given RH and corresponding water uptake (denoted as $f_{Mie}(RH)$). $f(RH)$ was approximated as a quadratic function of RH (denoted as $f_{app}(RH)$). The obtained approximated $f(RH)$ or $f_{app}(RH)$ values were compared with the values based on the Mie theory and showed a reasonable agreement between them. To the best of our knowledge, this is the first study that parametrizes the size dependency of $f(RH)$ for log normally distributed aerosols.

1. Introduction

Light absorption and scattering are the main processes involved in the interaction between aerosol particles and solar radiation (Ghan et al., 2012; Leibensperger et al., 2012). One of the essential aerosol characteristics in terms of interaction is hygroscopicity. The hygroscopic growth of aerosols is important in the interpretation of aerosol variables, such as visibility and aerosol optical depth (AOD) (Cheng et al., 2008; Liu et al., 2020). Hygroscopicity characterizes the ability of aerosols to take up water under subsaturated and supersaturated conditions and is a key parameter in determining the direct and indirect effects of aerosols on climate (Ogawa et al., 2016; Swietlicki et al., 2008; IPCC, 2013; Zhang et al., 2014). At an elevated ambient relative humidity (RH), hygroscopic aerosols have larger scattering cross sections than non-hygroscopic aerosols (Twohy et al., 2009). The hygroscopic growth of aerosols is specified by the growth factor (GF). The GF is defined to be the ratio of the wet particle radius to the dry particle radius, and can be

* Corresponding author.

E-mail address: jch@kiwu.ac.kr (C.H. Jung).

calculated as a function of RH (Eichler et al., 2008; Kreidenweis et al., 2008; Sorooshian et al., 2008):

$$GF(RH) = \frac{d_{wet}(RH)}{d_{dry}} = \left(\frac{V_{wet}(RH)}{V_{dry}} \right)^{1/3} = \left(\frac{V_{dry} + V_{water}(RH)}{V_{dry}} \right)^{1/3} = \left(1 + \frac{V_{water}(RH)}{V_{dry}} \right)^{1/3} \quad (1)$$

where, d_{wet} , d_{dry} , V_{wet} , V_{dry} , and V_{water} are the diameters and volumes of wet and dry aerosols and water, respectively.

To quantify the effect of hygroscopic growth on aerosol scattering, the scattering enhancement factor as a function of RH, which is $f(RH)$, is defined as follows (Lagrosas et al., 2019; Pitchford et al., 2007; Zhang et al., 2014; Zieger et al., 2010):

$$f(RH) = \frac{b_{ext}(RH)}{b_{ext}(dry)} \quad (2)$$

where $b_{ext}(RH)$ and $b_{ext}(dry)$ are the aerosol light extinction coefficients under enhanced RH conditions and dry conditions, respectively, and

$$b_{ext} = \int Q_{ext}(d_p) \frac{\pi}{4} d_p^2 n(d_p) dd_p \quad (3)$$

In Eq. (3), Q_{ext} is the single particle extinction efficiency and d_p is the particle diameter. Quantification of $f(RH)$ is critical to determine the response of aerosol optical properties to hygroscopic growth at various ambient RH and model the direct aerosol effect. Many previous studies have investigated aerosol hygroscopicity and associated growth through measurements and theoretical modeling (Liu et al., 2013; Zieger et al., 2010; 2011; 2013; 2014). These studies have mainly focused on the relationships among optical properties (such as aerosol scattering coefficients, radiative forcing, visibility), chemical species, and RH (Eichler et al., 2008; Kuang et al., 2016; Liu et al., 2008, 2013; Malm & Day, 2001; Malm et al., 2005; Cheng et al., 2008, 2003; Pan et al., 2009). There have also been many studies on the parameterization of these relations and the associated aerosol hygroscopicity, describing the variation in $f(RH)$ with RH (Carrico et al., 2003; Chen et al., 2014; Deng et al., 2016; Hänel, 1981; Im et al., 2001; Sheridan et al., 2002; Kasten, 1969; Kiehl et al., 2000; Kotchenruther & Hobbs, 1998; Kreidenweis et al., 2005; Malm et al., 1994; Petters & Kreidenweis, 2007; Randriamiarisoa et al., 2006; Rissler et al., 2006; Titos et al., 2016; Xu et al., 2002; Yan et al., 2009; Zhang et al., 2015; Zieger et al., 2011). Titos et al. (2016) reviewed several empirical parameterizations for the relationship between $f(RH)$ and RH and tested them using experimental data for hygroscopic growth. They also discussed the potential error sources in $f(RH)$ and found the error to be around 20–40% for moderately hygroscopic aerosols. Most of these previous parameterizations are based on empirically fitted functions.

Generally, observational $f(RH)$ can be obtained by measuring dry and wet scattering efficiencies using an optical instrument, such as a nephelometer (Cheng et al., 2008; Zhang et al., 2008). A theoretical $f(RH)$ can be estimated from dry and wet scattering efficiencies based on the Mie theory (Tao et al., 2017; Liu et al., 2008; Zieger et al., 2010; 2011). The Interagency Monitoring of Protected Visual Environments (IMPROVE) network measures scattering coefficients using nephelometry (Malm et al., 1994). One of the widely used expressions for the extinction coefficient, as a function of the aerosol mass concentration and $f(RH)$, is the reconstructed equation proposed by Malm et al. (1994), of which there are many different versions (Hand et al., 2019; Hand & Malm, 2007; Malm & Day, 2001; Malm et al., 1994; Malm et al., 2000). According to Malm et al. (1994), a scattering and absorption coefficient can be expressed as:

$$b_{scat} = 3.0 \{ C_{(NH_4)_2SO_4} + C_{NH_4NO_3} \} f_{AS}(RH) + 4.0 C_{OMC} + C_{soil} + 0.6 C_{CM} + 1.37 f_{ss}(RH) C_{ss}, \quad b_{abs} = 10.0 C_{BC} \quad (4)$$

where, b_{scat} and b_{abs} (in Mm^{-1}) are the reconstructed scattering and absorption coefficients, and $C_{(NH_4)_2SO_4}$, $C_{NH_4NO_3}$, C_{OMC} , C_{soil} , C_{CM} , C_{ss} , C_{BC} denote the mass concentrations of ammonium sulfate, ammonium nitrate, organic aerosol, soil, coarse mode particles, NaCl, and black carbon (BC) respectively. Here, the unit of mass concentration is $\mu g/m^3$, $f_{AS}(RH)$ denotes the scattering enhancement factor of AS (ammonium sulfate and ammonium nitrate), and $f_{ss}(RH)$ denotes that of NaCl. It should be noted that sulfate and nitrate aerosols have the same $f_{AS}(RH)$ in Malm's equation (Malm et al., 1994). Malm's reconstructed equation assumes a single $f_{AS}(RH)$ or $f_{ss}(RH)$, which is independent of particle size.

As discussed, one of the main drawbacks of Malm's equations is that they are unable to consider the size-dependent $f(RH)$, which is an important aspect of aerosol hygroscopicity (Pitchford et al., 2007; Chen et al., 2014; Laskina et al., 2015; Wang et al., 2018). For a given RH and chemical composition, different aerosol size distributions should yield different growth rates. Thus, $f_{AS}(RH)$ and $f_{ss}(RH)$, which comprise $f(RH)$ in Eq. (4) should be expressed as a function of particle size as well as particle composition and refractive index. Most previous studies, however, did not consider size-dependent effects due to the complexity in reflecting polydispersity and assumed $f(RH)$ to be constant at varying aerosol diameters (Molnár et al., 2020; Pitchford et al., 2007). However, many theoretical and empirical studies prove a size dependency and suggest that individual chemical substances should be specifically assessed with respect to particle size (Eichler et al., 2008; Liu et al., 2013; Chen et al., 2014; Wang et al., 2018; Qi et al., 2018; Shen et al., 2018; Zhang et al., 2018). From a theoretical point of view, when a size-dependent $f(RH)$ is to be considered, $f(RH)$ should be recalculated for each RH using a thermodynamic model and the Mie theory for each size distribution. Yet, such a recalculation is very complicated and, thus, instead of adopting this theoretical approach, simple parameterizations of hygroscopic growth factors for polydisperse aerosols need to be developed.

To consider size-dependent effects, Pitchford et al. (2007) revised the IMPROVE equation for fine- and accumulation-mode particles using the Mie theory. They used a bimodal aerosol size distribution with geometric mean diameters of 0.2 and 0.5 μm (geometric

standard deviation [GSDs] of 2.2 and 1.5, respectively) for fine and accumulation modes, respectively. Although their study used different $f(RH)$ for fine and accumulation modes (Pitchford et al., 2007), it did not consider the varying polydispersed size and changes in the size distribution.

In this study, an analytical approach to the approximated formula for the $f(RH)$ of the polydisperse size distribution of AS and NaCl was developed. For this development, we assumed that aerosols follow the lognormal size distribution (Jung et al., 2018, 2019; Seinfeld & Pandis, 1998).

In the following, the calculation of $f(RH)$ using the Mie theory (denoted as $f_{Mie}(RH)$) and the procedure of parameterizing $f(RH)$ using approximation (denoted as $f_{app}(RH)$) are described. Moreover, the parameterized and calculated $f(RH)$ values are compared and discussed. This new trial to parameterize the size dependency of $f(RH)$ enables the investigation of aerosol hygroscopic properties, such as AOD, visibility, and other related optical properties.

2. Materials and methods

2.1. Calculating $f(RH)$ using the Mie model ($f_{Mie}(RH)$)

Two step approach was used to obtain analytic approximations of the scattering enhancement factor: 1) calculating $f(RH)$ using the Mie theory ($f_{Mie}(RH)$), and 2) estimating $f(RH)$ using approximation ($f_{app}(RH)$) through multilinear regressions (MLR). Fig. 1 shows the procedures of those two steps. Spherical particles with an external mixture were assumed for the calculation under the particle size distribution and the complex particle refractive index at a wavelength of 0.55 μm (Malm et al., 1994). Here, the refractive index of AS and NaCl is 1.53 and 1.55, respectively (Malm, 2016; Pitchford et al., 2007).

$b_{ext}(dry)$ was calculated using the Mie theory with the dry size distribution and refractive index. $b_{ext}(RH)$ was calculated by applying GF(RH) to the size distribution using the Mie theory, which accounts for the change in particle volume and refractive index from the addition of water.

A dry log-normal size distribution with GSDs of 1.5, 2.0, and 2.5 was considered in order to consider broad ranges of aerosol size distribution (Ghan & Zaveri, 2007; Jung et al., 2019; Musante et al., 2002). The GF(RH) and water contents were calculated based on the Aerosol Inorganics Model (AIM), which is a thermodynamic equilibrium model with the “no solids” option of pure AS and NaCl (Clegg et al., 2003). A more detailed description of the Mie-based calculation of $f(RH)$ can be found in related studies (Zieger et al., 2010; 2014; van de Hulst, 1981; Lin et al., 2014). Hysteresis is the deliquescence branch with an abrupt increase in the water volume at the deliquescence RH (DRH), and the efflorescence branch with an abrupt evaporation at the crystallization relative humidity (CRH) (Pitchford et al., 2007; Randriamiarisoa et al., 2006). According to Pitchford et al. (2007), deliquescence is rarely observed in the

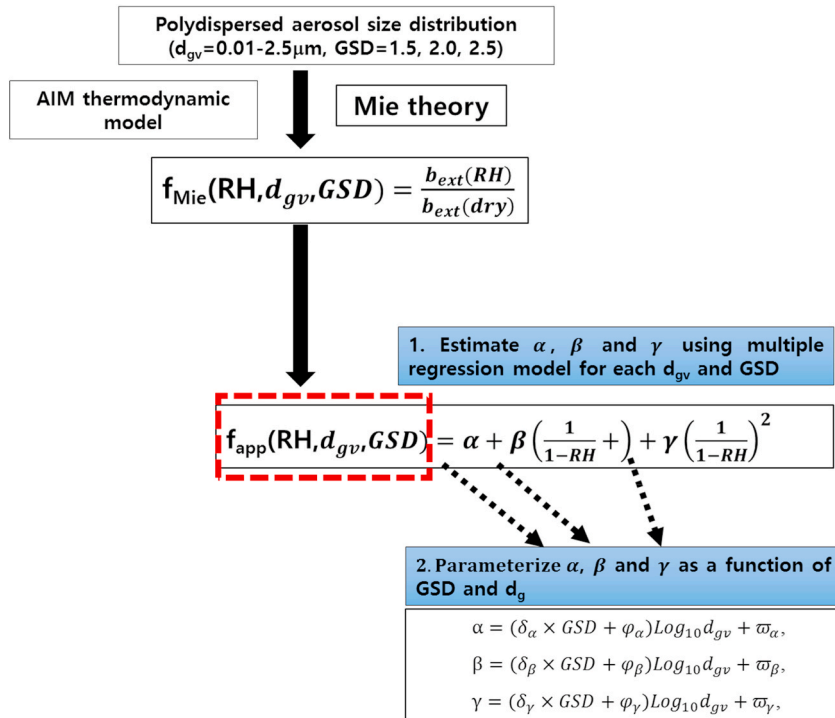


Fig. 1. Schematic of processes for obtaining the analytic expressions of the aerosol mass concentration and scattering enhancement factor, $f(RH)$, of polydispersed aerosols. GSD represents the geometric standard deviation, RH is the relative humidity, and AIM stands for aerosol thermodynamic model.

environment. We used the efflorescence growth curve for AS and NaCl. Because pure AS crystallizes at 37% RH, it is assumed that there is no hygroscopic growth, and that the $f(RH)$ is 1.0 below this RH. The water absorption curves for sea salt were nearly identical to those for NaCl. Below the crystallization point (i.e., RH 47% for NaCl), the growth factor was set to 1.0. Based on the growth curve, $f(RH)$ was introduced to illustrate the variability of the bulk scattering properties of aerosols at ambient RH, relative to a dry condition (Covert et al., 1972; Zieger et al., 2010).

2.2. Parameterization of $f(RH)$ using approximation ($f_{app}(RH)$)

Table 1 shows a comparison of the different types of parameterized expressions for $f(RH)$ (Kasten, 1969; Kotchenruther & Hobbs, 1998; Malm et al., 1994; Day et al., 2000; Kiehl et al., 2000; Im et al., 2001; Sheridan et al., 2001; Xu et al., 2002; Carrico et al., 2003; Liu et al., 2008, 2009; Yan et al., 2009; Pan et al., 2009; Zieger et al., 2011; Zhang et al., 2015; Chen et al., 2014; Deng et al., 2016).

IMPROVE uses approach by Malm et al. (1994). In this study, we approximated $f_{Mie}(RH)$ as a quadratic function of $\left(\frac{1}{1-RH}\right)$ for AS and NaCl (Malm et al., 1994; Deng et al., 2016).

$$f_{app,AS}(RH) = \alpha_{AS} + \beta_{AS} \left(\frac{1}{1-RH} \right) + \gamma_{AS} \left(\frac{1}{1-RH} \right)^2 \quad (5a)$$

$$f_{app,ss}(RH) = \alpha_{ss} + \beta_{ss} \left(\frac{1}{1-RH} \right) + \gamma_{ss} \left(\frac{1}{1-RH} \right)^2 \quad (5b)$$

$f_{Mie}(RH)$ was calculated using the Mie theory, with different aerosol size distributions. We approximated $f_{Mie}(RH)$ as linear combinations of $\left(\frac{1}{1-RH}\right)$ and $\left(\frac{1}{1-RH}\right)^2$ using multilinear regression (MLR). MLR is a statistical technique that uses several independent (explanatory) variables to predict the outcome of a dependent variable.

Here, the dependent variable is $f_{Mie}(RH)$ and the independent variables are $\left(\frac{1}{1-RH}\right)$ and $\left(\frac{1}{1-RH}\right)^2$.

α is the intercept, β , and γ are slope coefficients for the explanatory variable, and the subscripts AS and ss denote AS and NaCl. Using α , β , and γ , we can obtain the approximated $f(RH)$ ($f_{app}(RH)$) from Eq. (5). The coefficients were obtained by applying MLR to the $f_{Mie}(RH)$ values computed for the range of aerosol size distributions. These coefficients were statistically significant at a 0.05 significance level ($p < 0.05$).

The estimated coefficients (α , β , and γ) vary with the aerosol size distribution. For a given GSD, α , β , and γ can be approximated by functions of the volume-mean diameter.

$$\gamma = \eta_{\gamma} \log_{10}(d_{gv} / d_{g0}) + \varpi_{\gamma} \quad (6)$$

Table 1

Parameterized expressions for the aerosol scattering enhancement factor, $f_{app}(RH)$.

$f(RH)$	empirical coefficients	References
$f_{app}(RH) = \exp \left[c_1 + \left(\frac{c_2}{RH + c_3} \right) + \left(\frac{c_4}{RH + c_5} \right) \right]$	c_1, c_2, c_3, c_4, c_5	Kiehl et al. (2000); Im et al. (2001)
$f_{app}(RH) = 1 + a \times RH^b$	a, b	Kotchenruther and Hobbs (1998); Carrico et al. (2003); Zhang et al. (2015)
$f_{app}(RH) = \left(\frac{1}{1-RH} \right)^{\gamma}$	γ	Kotchenruther and Hobbs (1998)
$f_{app}(RH) = a \left(\frac{1}{1-RH} \right)^{\gamma}$	γ	Kasten (1969); Carrico et al. (2003); Zieger et al. (2011); Xu et al. (2002); Yan et al. (2009); Liu et al. (2008, 2009); Pan et al. (2009)
$f_{app}(RH) = \left(\frac{1-RH}{1-RH_{ref}} \right)^{\gamma}$	γ	Sheridan et al. (2001)
$f_{app}(RH) = a \left(\frac{1}{1-RH} \right)^{b \times RH}$	a, b	Chen et al. (2014)
$f_{app}(RH) = (1 + 2.42 \times RH^{4.67}) + 0.062 \times (1 - RH)^{-0.82}$		Liu et al. (2009)
$f_{app}(RH) = \left(1 + \kappa \frac{RH}{1-RH} \right)$	κ	Brock et al. (2016)
$f_{app}(RH) = \alpha + \beta \left(\frac{1}{1-RH} \right) + \gamma \left(\frac{1}{1-RH} \right)^2$	α, β, γ	Malm et al. (1994); Deng et al. (2016); Present study
$f_{app}(RH) = \alpha + \beta RH + \gamma RH^2$	α, β, γ	Kreidenweis et al. (2005); Ghan and Zaveri (2007)
$f_{app}(RH) = \frac{a + bRH}{1 + cRH}$	a, b, c	Day et al. (2000)

where, η_α , η_β , η_γ , ϖ_α , ϖ_β , and ϖ_γ are coefficients for α , β , and γ , d_{gv} is the volume-mean diameter in μm , and d_{g0} is the unit diameter ($=1 \mu\text{m}$). Here, all the logarithms of dimensioned quantities were standardized to avoid confusion with units (Matta et al., 2011).

Fig. 2 shows the coefficients for $f_{AS}(RH)$ as a function of the $\log_{10}d_{gv}$. GSDs of 1.5, 2.0, and 2.5 were considered. For a small aerosol-size range ($d_{gv} < 0.03 \mu\text{m}$), the coefficients α , β , and γ converge to 0.3, 0.83, and 0.07, respectively. Meanwhile, for large particles that are $>1 \mu\text{m}$ in volume-mean diameter ($d_{gv} > 1 \mu\text{m}$), the coefficients α , β , and γ converge to 0.85, 0.28, and 0.0, respectively. In these cases, η_α , η_β , and η_γ should be 0 in Eq. (6). An intermediate size exists between the two size ranges ($0.03 \mu\text{m} < d_{gv} < 1 \mu\text{m}$); the approximations for this intermediate size vary with size, meaning that they should be expressed as d_{gv} and GSD, as shown in Fig. 2. Coefficients for the intermediate size can be approximated by the linear function $\log_{10}(d_{gv}/d_{g0})$, as described in Eq. (6). For example, the intercept coefficient α_{AS} (α for AS) can be approximated to be $0.49\log_{10}(d_{gv}/d_{g0})+0.84$, $0.35\log_{10}(d_{gv}/d_{g0})+0.84$, and $0.22\log_{10}(d_{gv}/d_{g0})+0.84$ for a GSD of 1.5, 2.0, and 2.5, respectively.

Fig. 3 shows the coefficients for $f_{ss}(RH)$. The calculated and approximated results for α_{ss} , β_{ss} , and γ_{ss} were compared for the GSDs of 1.5, 2.0, and 2.5, respectively. α_{ss} , β_{ss} , and γ_{ss} can be approximated by the linear function $\log_{10}(d_{gv}/d_{g0})$ (Fig. 3). For example, α_{ss} can be approximated to be $1.3\log_{10}(d_{gv}/d_{g0})+1.4$ and $0.88\log_{10}(d_{gv}/d_{g0})+1.4$ for GSDs of 1.5 and 2.0 in the intermediate sizes ($0.03 \mu\text{m} < d_{gv} < 1 \mu\text{m}$), respectively, and converges to 0.0 and 1.35 for small ($d_{gv} < 0.03 \mu\text{m}$) and large particles ($d_{gv} > 1 \mu\text{m}$), respectively.

To consider polydispersity with different GSDs, we required more parameterizations for the coefficients η_α , η_β , and η_γ in Eq. (6) in the intermediate-size range. Here η_α , η_β , and η_γ can be estimated by linear fitting as a function of the GSD with the slope (δ) and intercept (ϕ). Finally, we approximated η_α , η_β , and η_γ again as a function of GSD, as shown in Eq. (7).

$$\eta_\alpha = \delta_\alpha \times GSD + \phi_\alpha$$

$$\eta_\beta = \delta_\beta \times GSD + \phi_\beta$$

$$\eta_\gamma = \delta_\gamma \times GSD + \phi_\gamma$$

(7)

Fig. 4 shows the approximated estimation of the slope coefficients (η_α , η_β , and η_γ) as a function of GSD for AS and NaCl. η_α , η_β , and η_γ for the intermediate sizes can be approximated as $(-0.27\text{GSD}+0.89)$, $(0.29\text{GSD}-0.94)$, and $(0.03\text{GSD}-0.085)$ for AS, and $(-0.68\text{GSD}+2.29)$, $(0.7\text{GSD}-2.77)$, and $(0.31\text{GSD}-0.91)$ for NaCl, respectively.

Then, the resultant coefficients for the polydispersed aerosol with $f_{app}(RH)$ were expressed as a function of the d_{gv} and GSD by

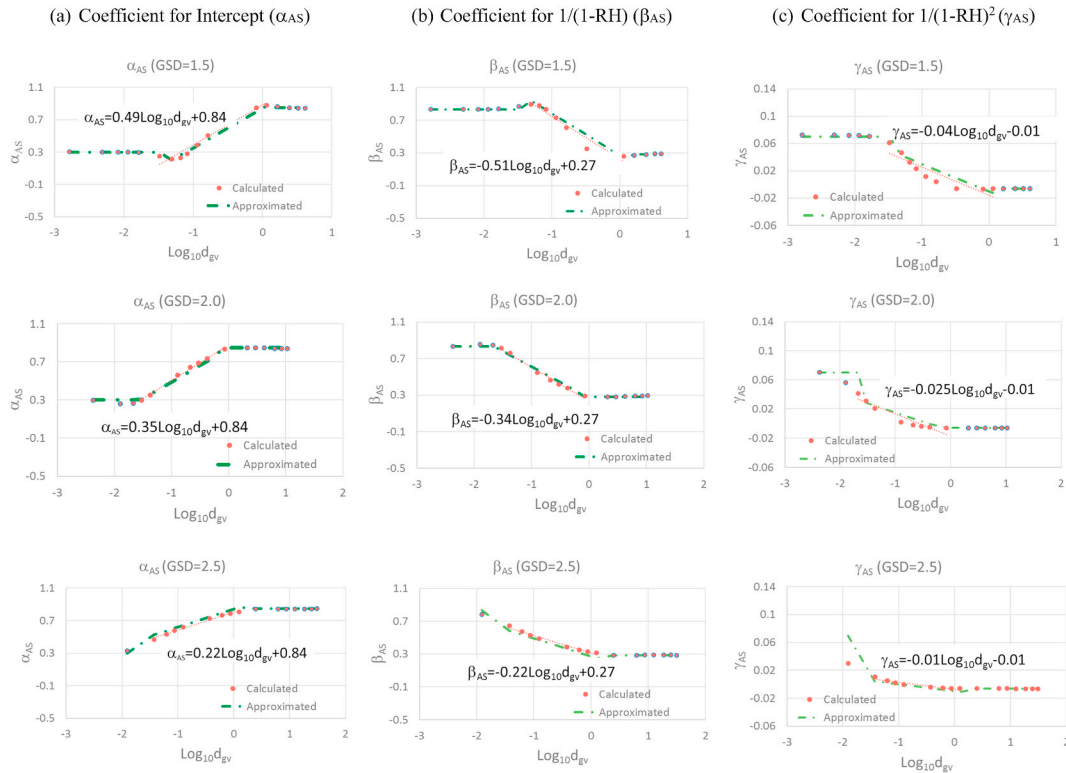


Fig. 2. Coefficients of the aerosol mass concentration and scattering enhancement factor, $f(RH)$, of AS (ammonium sulfate and ammonium nitrate aerosols) as a function of the volume-mean diameter (d_{gv}) and geometric standard deviation (GSD). Circles are calculated coefficients and dotted and dash-dot lines are approximated coefficients.

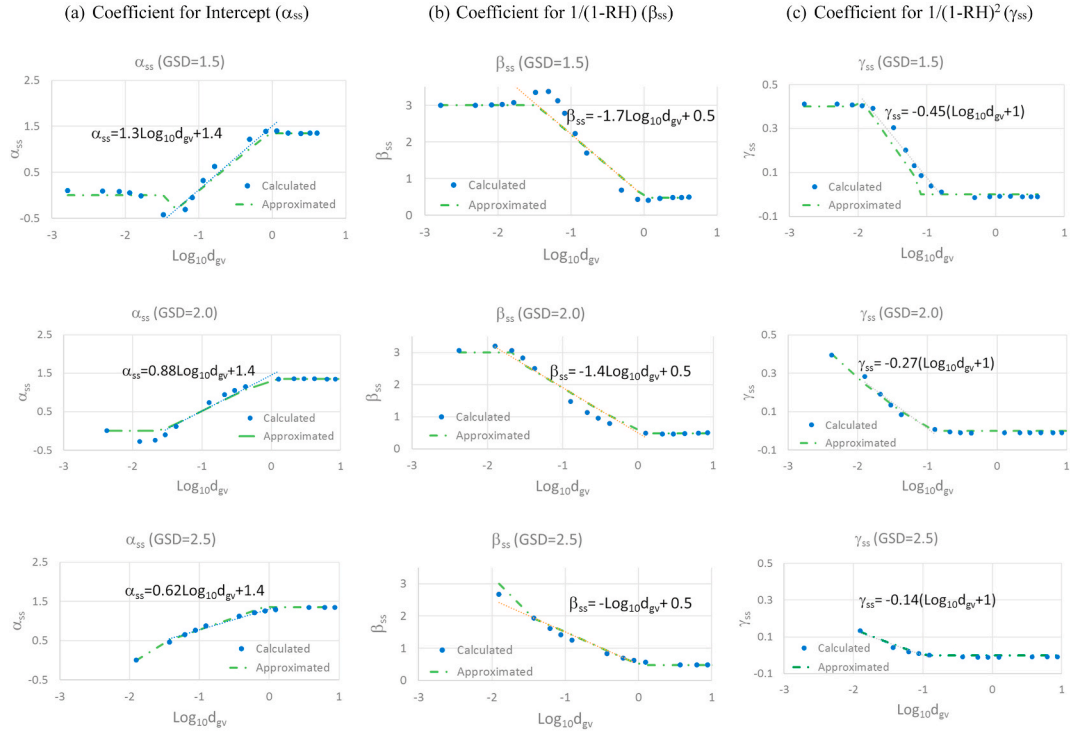


Fig. 3. Coefficients of the aerosol mass concentration and scattering enhancement factor, $f(RH)$, of NaCl as a function of the volume-mean diameter (d_{gv}) and geometric standard deviation (GSD). Circles are calculated coefficients and dotted and dash-dot lines are approximated coefficients.

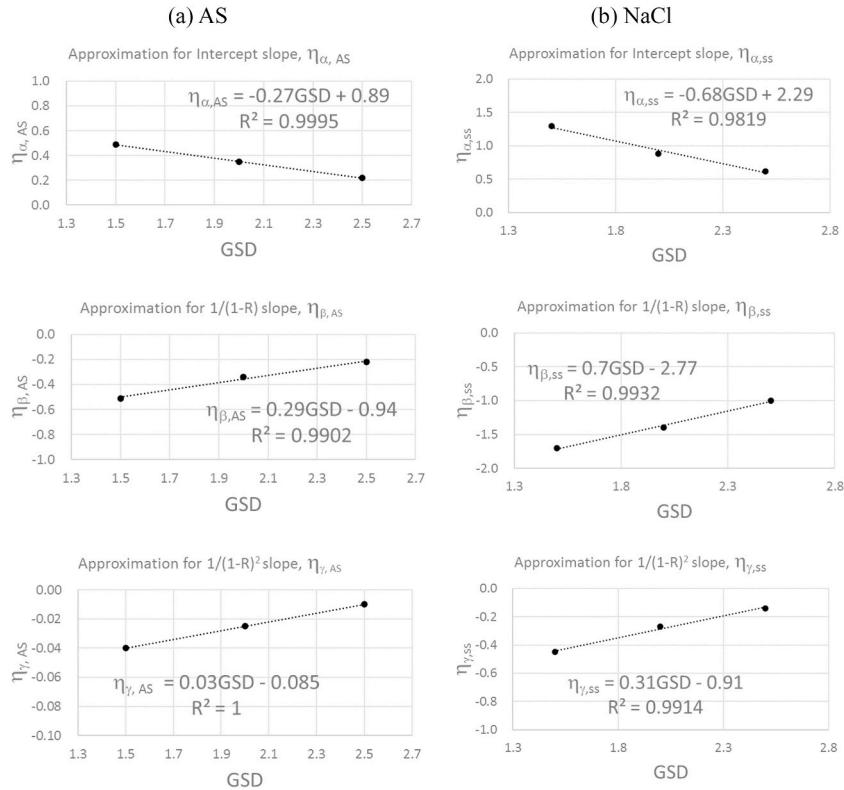


Fig. 4. Approximated estimation for the slope of each of the coefficients (α , β , γ) as a function of GSD for AS and NaCl.

combining Eqs. (6) and (7) as follows:

$$\begin{aligned}\alpha &= (\delta_\alpha \times GSD + \phi_\alpha) \text{Log}_{10}(d_{gv}/d_{g0}) + \varpi_\alpha \\ \beta &= (\delta_\beta \times GSD + \phi_\beta) \text{Log}_{10}(d_{gv}/d_{g0}) + \varpi_\beta \\ \gamma &= (\delta_\gamma \times GSD + \phi_\gamma) \text{Log}_{10}(d_{gv}/d_{g0}) + \varpi_\gamma\end{aligned}\quad (8)$$

The approximated subsequent expressions and detailed coefficients for each d_{gv} , d_g , and GSD are summarized in Tables 2 and 3 for AS and NaCl, respectively. It has to be noted that the parameterization in this study is based on d_{gv} . However, d_{gv} can be converted to the number-mean diameter (d_g). The relationship between d_g and d_{gv} can be expressed as follows (Hinds, 1999).

$$d_{gv} = d_g \exp[3 \ln^2(GSD)] \quad (9)$$

While d_{gv} is useful for volume-size distributions that are, in turn, useful when dealing with air quality and optical properties of aerosols, d_g is useful for number-size distributions that are, in turn, useful when dealing with microphysical processes.

3. Results and discussion

3.1. Comparison of approximated results with those from the Mie theory

Fig. 5 shows $f(\text{RH})$ for different size distributions with different GSDs and d_g for AS, with an RH of 37–95%. Fig. 5(a) shows the $f_{\text{Mie}}(\text{RH})$ with a GSD of 1.5 and different d_g of 0.05, 0.1, 0.2, and 0.7 μm . As shown in Fig. 5(a), $f_{\text{Mie}}(\text{RH})$ decreases as d_g increases. Fig. 5(a) also shows that the original $f(\text{RH})$ from Malm et al. (1994), which was parameterized based on the data published by Tang et al. (1981), is between the $f(\text{RH})$ values varying with d_g for a given RH and GSD.

Fig. 5(b) shows the $f_{\text{Mie}}(\text{RH})$ with the different GSDs of 1.5 and 2.5. $f_{\text{Mie}}(\text{RH})$ converges as the GSD increases, regardless of particle size. For a GSD of 2.5, the $f(\text{RH})$ with a d_g of 0.1 μm is similar to that with a d_g of 0.7 μm . Fig. 5(c) and (d) show a comparison between the Mie-calculated ($f_{\text{Mie}}(\text{RH})$) and approximated $f(\text{RH})$ ($f_{\text{app}}(\text{RH})$). Fig. 5(c) and (d) show the comparison between the d_g of 0.1 and 0.7 μm for GSDs of 1.5 and 2.5, respectively. The Mie calculated and approximated results show a good agreement. The mean absolute errors (MAE) were 13.07% for a d_g of 0.1 μm with a GSD of 1.5, 9.97% for a d_g of 0.7 μm with a GSD of 1.5, 12.6% for a d_g of 0.1 μm with a GSD of 2.5, and 6.17% for a d_g of 0.7 μm with a GSD of 1.5 over the RH range of 37–95%. MAE represents the average of the absolute differences between prediction and actual observation across the test samples, where all individual differences are equally weighted. It measures the average magnitude of errors in a set of predictions, without considering their direction (Habyarimana et al., 2019). Here, MAE is the average of the absolute difference over the RH range between the approximated and Mie calculated results.

$$\text{MAE}(a-b) = \frac{1}{n} \sum_{\text{RH}=a}^b \frac{|X_{\text{Mie,RH}} - X_{\text{app,RH}}|}{X_{\text{Mie,RH}}} \quad (10)$$

where, $X_{\text{Mie,RH}}$ and $X_{\text{app,RH}}$ are the Mie calculated and approximated results at a given RH, respectively.

In the case of AS in the RH range of 37–95%, MAE ranges from 2.03 to 20.16%. For a RH range of 37–90%, MAE ranges from 1.96 to 15.44%.

Along with AS, one of the important factors contributing to regional differences in aerosols in the model is NaCl. The original Malm's reconstructed method included NaCl aerosols in the coarse mode (Malm et al., 1994). Although the total NaCl surface mass concentration is dominated by super-micron particles, solar radiation scattering is more efficient on particle diameters between 0.2 and 1.0 μm as compared to other diameters (Ayash et al., 2008). Sub-micron sea-salt aerosols account for ~20% of the total light scattering (Ayash et al., 2008; Quinn et al., 1996). For this reason, $f(\text{RH})$ for NaCl is important for sub-micron particles as well as for super-micron particles. In this study, we parameterize the $f(\text{RH})$ of NaCl polydispersed aerosols, which include fine and accumulation modes.

Fig. 6(a) shows an $f_{\text{Mie}}(\text{RH})$ for NaCl in the range of 47–95% and a different d_g of 0.1, 0.2, 0.7, and 2.5 μm . The $f_{\text{Mie}}(\text{RH})$ in Fig. 6(a) is for a GSD of 1.5. $f(\text{RH})$ decreases as d_g increases (Fig. 6). Fig. 6(b) shows $f_{\text{Mie}}(\text{RH})$ with different GSDs of 1.5 and 2.5 and a d_g of 0.1, 0.7, and 2.5 μm for NaCl. $f_{\text{Mie}}(\text{RH})$ converges as the GSD and d_g increase. Fig. 6(c) and (d) show a comparison between the Mie calculated ($f_{\text{Mie}}(\text{RH})$) and approximated $f(\text{RH})$ ($f_{\text{app}}(\text{RH})$) in the case of NaCl. Comparisons are made between a d_g of 0.1 and 2.5 μm for a GSD of 1.5 and 2.5. Fig. 6(c) and (d) show that the Mie calculated and approximated $f(\text{RH})$ are in good agreement. In the case of NaCl,

Table 2

Approximated coefficients (α , β , γ) for the hygroscopic growth factor of AS ($f_{\text{app,AS}}(\text{RH}) = \alpha_{\text{AS}} + \beta_{\text{AS}} \left(\frac{1}{1-\text{RH}} \right) + \gamma_{\text{AS}} \left(\frac{1}{1-\text{RH}} \right)^2$).

Size range	coef. Intercept, α_{AS}	coef. $1/(1-\text{RH})$, β_{AS}	coef. $1/(1-\text{RH})^2$, γ_{AS}
$d_{gv} < 0.3 \mu\text{m}$	0.3	0.83	0.07
$0.3 \mu\text{m} < d_{gv} < 1 \mu\text{m}$	$(-0.27\text{GSD} + 0.89)\text{Log}_{10}(d_{gv}/d_{g0}) + 0.84$	$(0.29\text{GSD} - 0.94)\text{Log}_{10}(d_{gv}/d_{g0}) + 0.27$	$(0.03\text{GSD} - 0.085)\text{Log}_{10}(d_{gv}/d_{g0}) - 0.01$
$d_{gv} > 1 \mu\text{m}$	0.85	0.28	0.0

Table 3

Approximated coefficients (α , β , γ) for the hygroscopic growth factor of NaCl $\left(f_{app, ss}(RH) = \alpha_{ss} + \beta_{ss} \left(\frac{1}{1-RH}\right) + \gamma_{ss} \left(\frac{1}{1-RH}\right)^2\right)$.

Size range	coef. Intercept, α_{ss}	coef. $1/(1-RH)$, β_{ss}	Range	coef. $1/(1-RH)^2$, γ_{ss}
$d_{gv} < 0.3 \mu m$	0.00	3.00	$d_{gv} < 0.01 \mu m$	0.40
$0.3 \mu m < d_{gv} < 1 \mu m$	$(-0.68GSD + 2.29)\text{Log}_{10}(d_{gv}/d_{g0}) + 1.4$	$(0.7GSD - 2.77)\text{Log}_{10}(d_{gv}/d_{g0}) + 0.5$	$0.01 \mu m < d_{gv} < 0.1 \mu m$	$(0.31GSD - 0.91)(\text{Log}_{10}(d_{gv}/d_{g0}) + 1)$
$d_{gv} > 1 \mu m$	1.35	0.48	$d_{gv} > 0.1 \mu m$	0.00

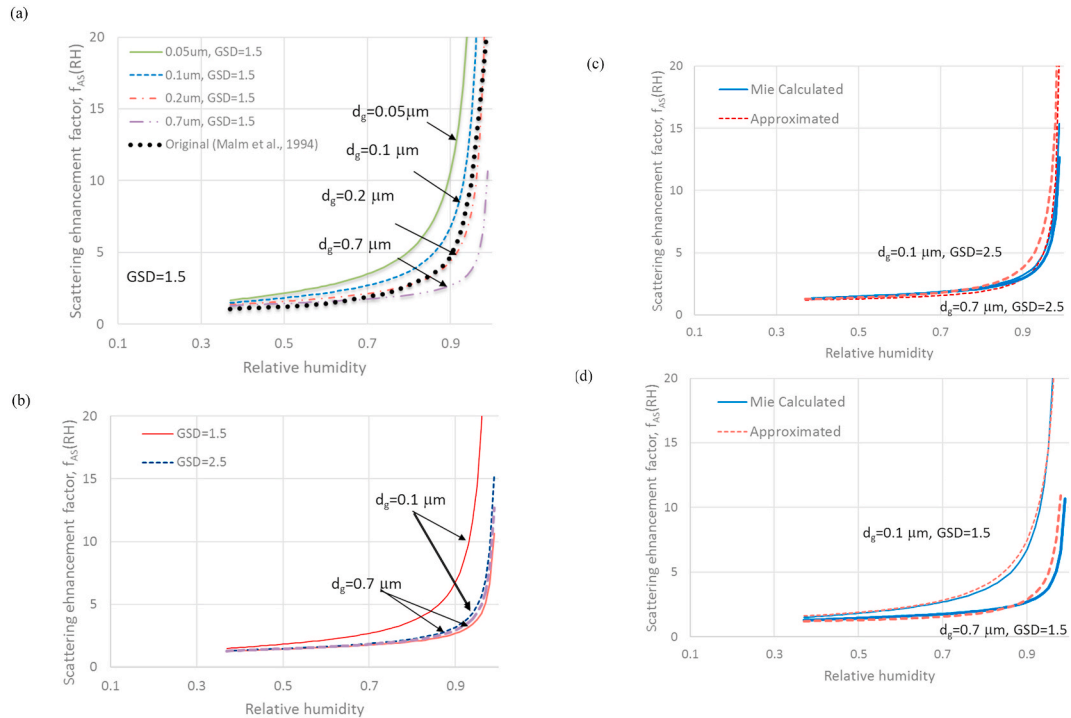


Fig. 5. Aerosol mass concentration and scattering enhancement factor, $f(RH)$, for different size distributions with different GSD and geometric mean diameters (d_g) for AS, with an RH of 37–95%.

MAE ranges from 4.48 to 27.66% for an RH ranging between 47 and 95% and from 3.63 to 27.04% for an RH ranging between 47 and 90%.

Table 4 summarizes MAEs for AS and NaCl. The average MAEs for 27 cases are 7.6% (RH of 37–95%, AS), 5.55% (RH of 37–90%, AS), 9.86% (RH of 37–95%, NaCl), and 8.03% (RH of 37–90%, NaCl).

3.2. Size and type dependency of $f(RH)$

Many previous studies have shown that different types of aerosols exhibit different hygroscopic, optical and microphysical behaviors, which are affected by aerosol physico-chemical characteristics and other factors, including location, time, aging process, mixing state, and source (Chen et al., 2014; Wang et al., 2018). For example, different types of aerosols, such as biomass-burning and marine aerosols, show different hygroscopic growth factors (Asmi et al., 2010). According to Titos et al. (2016), the highest $f(RH)$ values were measured in clean marine environments, where pollution had a major influence on $f(RH)$. Dust aerosols tend to have the lowest reported hygroscopicity among the aerosol types studied (Titos et al., 2016).

Additional information is required for understanding the size dependency of aerosol hygroscopicity (Chen et al., 2014, 2019; Eichler et al., 2008; Sorooshian et al., 2008; Wang et al., 2018). Different aerosol types tend to have different sizes, depending on their composition and sources. Moreover, although aerosol types are identical, different sizes of particles can undergo different size-dependent processes under ambient conditions, such as coagulation, condensation, and deposition, which results in different hygroscopic growth behaviors. Thus, further experimental and theoretical studies on the particle size-resolved hygroscopic behavior of aerosols are required to understand the mechanisms controlling the evolution of aerosol pollution (Brock et al., 2016; Kuang et al., 2018; Wang et al., 2018).

In this study, the approximated coefficients for both AS and NaCl show that there are sub-micron particle-size ranges that show a

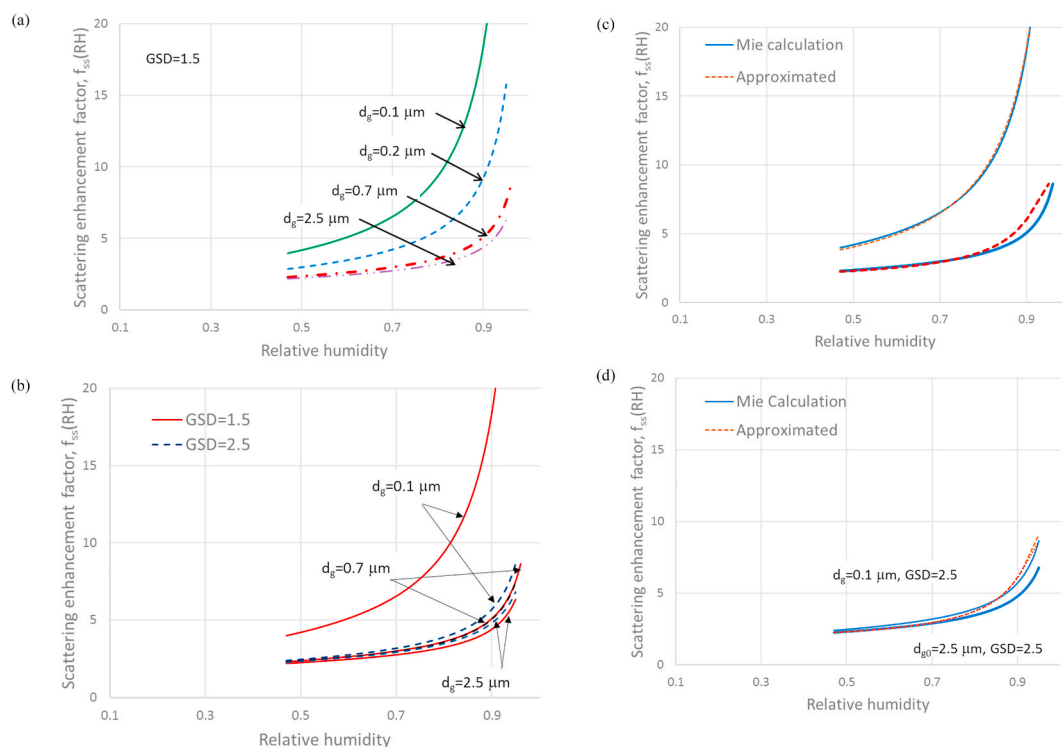


Fig. 6. Aerosol mass concentration and scattering enhancement factor, $f_s(RH)$, for different size distributions with different GSD and geometric mean diameters (d_g) for NaCl, with an RH of 47–95%.

Table 4

Mean absolute errors (MAE, %) for different size distributions.

			AS		NaCl	
d_g (μm)	d_{gv} (μm)	GSD	MAE (37–95%)	MAE (37–90%)	MAE (47–95%)	MAE (47–90%)
0.01	0.016	1.5	4.76	5.04	7.15	7.12
0.02	0.033		2.04	1.96	14.52	13.83
0.05	0.082		4.29	3.46	27.66	27.04
0.10	0.164		13.07	9.94	11.31	11.15
0.20	0.328		20.17	15.44	10.48	7.50
0.50	0.819		5.10	4.04	10.16	6.85
0.70	1.146		9.97	5.88	15.94	11.26
1.00	1.638	2.0	8.49	5.16	9.60	5.96
2.50	4.094		7.45	4.78	7.31	4.38
0.01	0.042		3.53	3.30	9.26	10.14
0.02	0.085		6.05	4.37	7.81	7.97
0.05	0.211		9.97	7.27	4.49	3.62
0.10	0.423		6.20	4.20	6.93	5.06
0.20	0.845		7.89	7.13	16.10	17.38
0.50	2.113	2.5	8.09	4.91	9.09	5.29
0.70	2.959		8.00	4.91	8.53	4.95
1.00	4.226		7.72	4.82	7.77	4.51
2.50	10.566		7.38	4.84	7.27	4.51
0.01	0.124		5.46	5.49	5.69	5.77
0.02	0.248		5.34	5.31	5.83	6.10
0.05	0.621		10.05	9.00	16.24	17.21
0.10	1.241	2.5	6.17	4.52	7.20	6.04
0.20	2.483		7.23	4.58	7.36	4.35
0.50	6.207		7.67	4.80	7.79	4.53
0.70	8.689		7.59	4.81	7.60	4.47
1.00	12.413		7.53	4.83	7.56	4.51
2.50	31.033		8.07	4.98	9.37	5.32

size dependency of these coefficients. For a sub-micron size range, for example, an increase in particle size would lead to a lower $f(\text{RH})$ for both AS and NaCl. Here, it should be noted that $f(\text{RH})$ is based on the assumption that aerosols are externally mixed.

For the overall $f(\text{RH})$ of internally mixed aerosols, however, there has been an ongoing discussion regarding the size dependency. If we consider internally mixed aerosols to be composed of hygroscopic and non-hygroscopic species, the overall $f(\text{RH})$ depends on the fraction of hygroscopic aerosol species. Generally, the fractions of aerosol species determine aerosol types and, thus, the $f(\text{RH})$ of internally mixed aerosols is type dependent. For example, $f(\text{RH})$ values are larger for marine sites compared to other environments due to the high hygroscopicity of NaCl (Titos et al., 2016; Zieger et al., 2011).

Aerosol mixtures are related to aerosol type and $f(\text{RH})$ depends on the fractions of hygroscopic species with different hygroscopic properties. If we calculate $f(\text{RH})$ for an internal mixing, also the volume fraction of the chemical composition will have an influence. For example, an increase in the organic fraction, which is known to be weakly hygroscopic (Malm et al., 2005; Wang et al., 2018), will lower the $f(\text{RH})$ of internally mixed aerosols.

Thus, the size dependency of $f(\text{RH})$ is related to the fraction of each aerosol and the degree of aerosol hygroscopicity (Zieger et al., 2013). Zieger et al. (2010) indicated that small and less hygroscopic aerosols can have the same $f(\text{RH})$ as large and more hygroscopic aerosols due to the compensation between size and hygroscopicity. According to Zieger et al. (2010), for example, the observed $f(\text{RH})$ did not show a clear seasonal trend and was not positively correlated with the occurrence of sea salt, as inferred from the volume fraction of large particles. This compensation can be seen in the Mie calculations for aerosols consisting of various fractions of highly and weakly hygroscopic inorganic aerosols (Zieger et al., 2010). Other studies (Sheridan et al., 2001; Titos et al., 2016) showed that a behavior as a function of size is more evident in clean than in polluted marine situations and that the difference between $f(\text{RH})$ for different size cuts was more obvious for clean marine conditions. From these previous studies, it can be inferred that the size dependency of $f(\text{RH})$ depends on many environment-related physico-chemical parameters, which make the quantification of $f(\text{RH})$ difficult. Clearly, more studies on this size dependency, using observations and experimental frameworks, are required.

4. Conclusions

In this study, we developed analytical expressions to approximate $f(\text{RH})$ for ammonium sulfate and sea salt aerosols. To obtain these values, we first computed mass growth factors based on aerosol thermodynamic models. Then, the optical properties of the aerosols were calculated using the Mie theory for a range of different aerosol properties and size distributions. From these obtained values, $f(\text{RH})$ was approximated for polydisperse aerosols as a quadratic function of RH and the coefficients were parameterized as a function of d_{gv} and GSD.

The results show that the obtained approximated coefficients and $f(\text{RH})$ correspond well with the Mie calculated values for the polydisperse particle sizes. Consequently, this study suggests a simple and accurate formula for $f(\text{RH})$. This formula estimates $f(\text{RH})$ and related optical properties, such as the aerosol extinction coefficient and AOD, with confidence.

The current approximated coefficients for $f(\text{RH})$ ($f_{\text{app}}(\text{RH})$) are based on a quadratic curve following Malm et al. (1994). There are many different formulas for the parameterization of $f(\text{RH})$; however, it should be noted that the previously proposed empirical expressions for $f(\text{RH})$ depend only on RH.

To the best of our knowledge, this study is the first to parametrize the size dependency of $f(\text{RH})$ for polydisperse aerosols. However, there remains a need for further investigation into more accurate, simple, and optimized methods, which will contribute towards improving hygroscopic studies. To consider size-resolved aerosol optical properties, such as extinction and scattering coefficients, mass scattering efficiency should be expressed as a function of size. We will perform a more intensive investigation into the parameterization of the size-dependent mass scattering efficiency in a follow-up study. It should also be noted that our parameterization covers the entire size range, including Aitken and accumulation modes, but neglects Kelvin effects, which needs to be improved when considering curvature effects.

There has been an ongoing discussion about the size dependency of $f(\text{RH})$ in the real atmosphere. $f(\text{RH})$ under real conditions depends on many factors, such as mixture- and hygroscopicity-related environmental and meteorological parameters. This study focused on aerosol sizes among those factors and, hence, to isolate the size dependency of $f(\text{RH})$, other factors have to be theoretically and experimentally confined.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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